# **Copper** Solvent Extraction

29

Cu

47

Copper 63.546

0

NI

16

Nickel

8.6934

- 10

**BASF** We create chemistry

12

Zn

48

Zinc

65.38

d

411

30

1.00

# **BASF LIX® technology –** recovery of copper from leach solutions

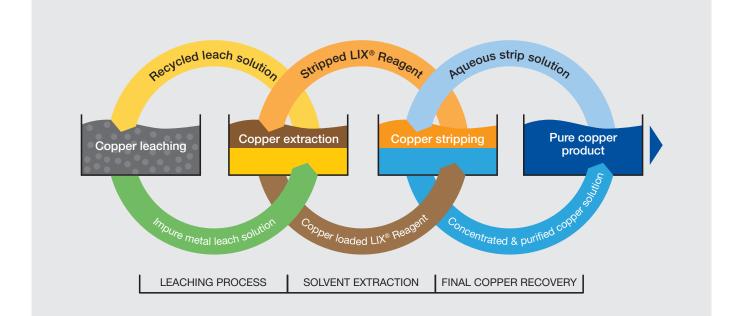
Leaching copper from oxide ores, low grade secondary sulfide ores, and primary sulfide concentrates with aqueous acid or ammonia solutions, followed by solvent extraction (SX) has been practiced commercially since the late 1960's. Over time, LIX<sup>®</sup> Reagents (hydroxyoximes – chelating agents) used in these solvent extraction processes have become more efficient, more selective and more economical to use.

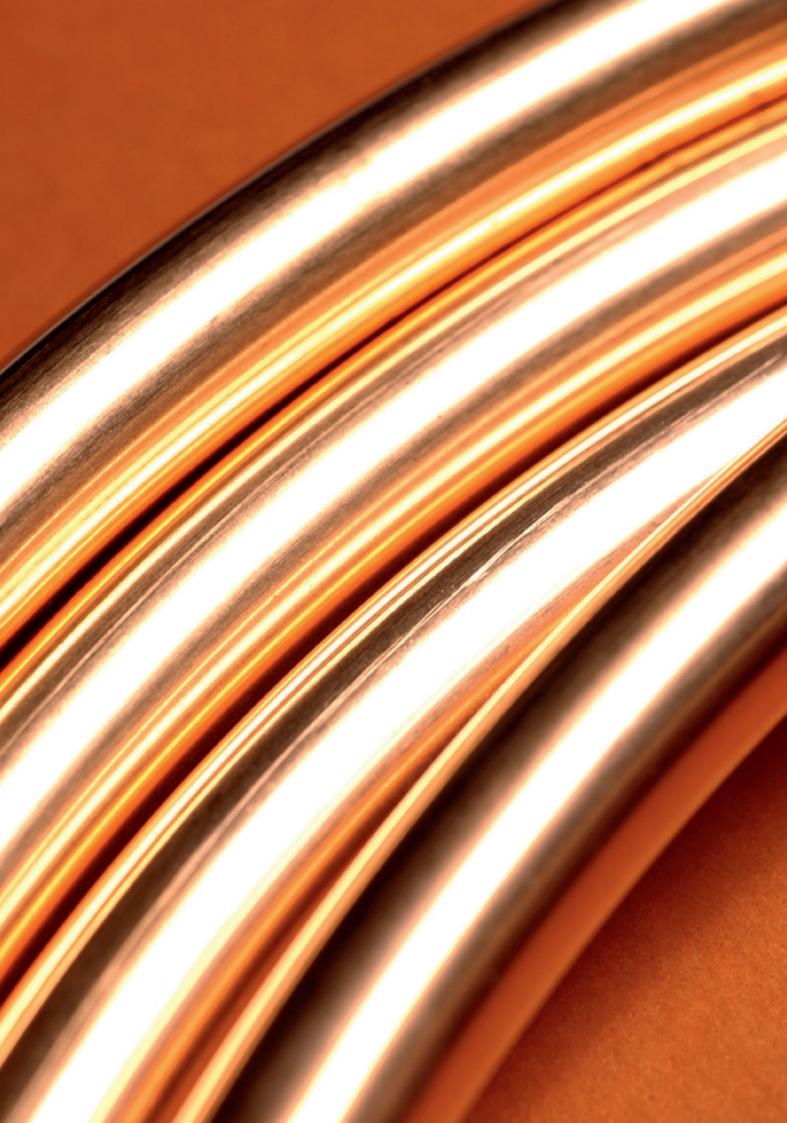
The benefits and objectives of the copper solvent extraction process are purification of copper, concentration of copper and conversion of the copper to a form that allows BASF's customers to produce desired final metal products including cathode, powder and crystals.

BASF produces a complete line of copper solvent extraction reagents, including ketoximes, aldoximes, non-modified blends and modified blends of ketoximes and aldoximes. These reagents and reagent blends allow BASF to supply its customers with an optimized reagent formulation tailored to meet the needs of their specific applications.

The overall leaching, solvent extraction and final metal production processes can be summarized in the following simplified flowsheet.

# LEACHING PROCESS/SOLVENT EXTRACTION/ FINAL COPPER RECOVERY





#### **Advantages of BASF LIX® Reagents**

- Fast kinetics: Rapid copper uptake and copper stripping
- Selective: Very high iron rejection
- Net copper transfer: Can be adjusted to maximize process performance
- Ease of operations: Cleaner phase separation and lower crud formation
- High purity final copper product: Lower entrainment levels and lower transfer of contaminants
- Low viscosity: Operate better at high reagent concentrations and low temperatures
- Lower costs: Lower losses, higher net transfer, improved chemical stability

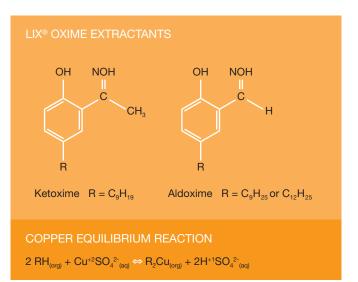
#### **Service**

All LIX<sup>®</sup> Reagents are backed by BASF's comprehensive MSP<sup>®</sup> Minesite Services Program and BASF's qualified experts and engineers.

# **Computer modeling**

The BASF Isocalc<sup>®</sup> computer simulation program provides plant operators with fast and accurate simulations of virtually any copper solvent extraction process involving sulfuric acid leaching followed by copper solvent extraction with LIX<sup>®</sup> Reagents. The Isocalc<sup>®</sup> program will generate extraction and stripping isotherms and construct McCabe-Thiele diagrams for a large number of circuit configurations. The program can also be calibrated to match the performance of commercial SX plants, which allows the production engineer to accurately determine the impact that changes in PLS chemistry, reagent concentration, reagent blend and/or circuit configuration will have on copper recovery.

# THE CHEMICAL STRUCTURES FOR THE KETOXIME AND ALDOXIME REAGENTS



**Ketoximes:** Moderately strong extractants and can be used without equilibrium modifiers.

**Aldoximes:** Very strong extractants and have to be used in mixture with an equilibrium modifier or ketoxime to aid the stripping process and obtain adequate net copper transfer.

LIX<sup>®</sup> Reagents work on a hydrogen ion cycle. Acid is generated in electrowinning, transferred to the organic in stripping in exchange for copper and then exchanged for copper in extraction.

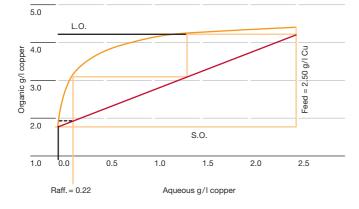


Final Copper Cathode product bundled, weighed and ready for shipment.

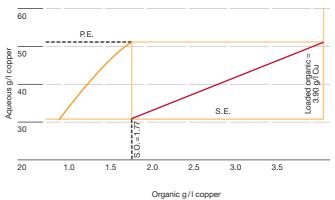




#### Equilibrium extraction isotherm/two stage McCabe-Thiele diagram with a typical LIX<sup>®</sup> Reagent



Equilibrium strip isotherm/one strip stage McCabe-Thiele diagram with a typical LIX<sup>®</sup> Reagent



SX plant configurations and final copper product

Commercial copper SX plants are designed around one of the several proven circuit configurations: series, series parallel, modified series parallel and optimum series parallel (or interlaced). The circuit configuration chosen will depend on several variables unique to the site including economics, leach solution chemistry and the operating philosophy of the company associated with each plant. The solvent extraction process can produce several different final copper products, however, most SX plants produce copper cathode.

#### Europe BASF SE

Mining Solutions Carl-Bosch-Str. 38 67056 Ludwigshafen Germany Phone: +49 621 60 0 Fax: +49 621 60 42525

#### **North America**

BASF Corporation 3231 E Valencia Road Tucson, AZ 85706 US Phone: +1 520 629 3208 Fax: +1 520 624 0912

### South America

BASF CHILE S.A. Av. Carrascal Nº 3851 Quinta Normal Santiago Chile Phone: +56 2 2640 7000 Fax: +56 2 775 3095

#### Africa

#### **BASF South Africa (Pty) Ltd.**

852 Sixteenth Road Midrand, P.O. Box 2801 Halfway House 1685 South Africa Phone: +27 11 203 2400 Fax: +27 11 203 2431

#### Australia

BASF Australia Ltd. Level 12, 28 Freshwater Place VIC 3006, Southbank Australia Phone: +613 8855 6600 Fax: +613 8855 6511

For further information: miningsolutions@basf.com www.mining-solutions.basf.com The descriptions, designs, data and information contained herein are presented in good faith, and are based on BASF's current knowledge and experience. They are provided for guidance only, and do not constitute the agreed contractual quality of the product or a part of BASF's terms and conditions of sale. Because many factors may affect processing or application/use of the product, BASF recommends that the reader carry out its own investigations and tests to determine the suitability of a product for its particular purpose prior to use. It is the responsibility of the recipient of product to ensure that any proprietary rights and existing laws and legislation are observed. No warranties of any kind, either express or implied, including, but not limited to, warranties of merchantability or fitness for a particular purpose, are made regarding products described or designs, data or information set forth herein, or that the products, descriptions, designs, data and information given in this publication may change without prior information. The descriptions, designs, data, and information furnished by BASF hereunder are given gratis and BASF assumes no obligation or liability for the descriptions, designs, data or information given or results obtained, all such being given and accepted at the reader's risk. (05/2021) (9) = registered trademark of BASF SE